L3 ANSWER 1 OF 6 CAPLUS COPYRIGHT 2003 ACS

AN 1998:142921 CAPLUS

TI Olefin trimerization with 1,3,5-triazacyclohexane complexes of chromium

STN 1-17-03 Author Search

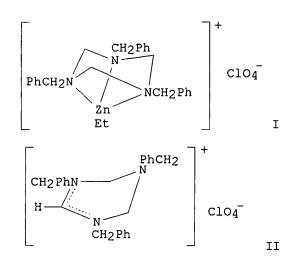
- AU Koehn, R. D.; Haufe, M.; Kociok-Koehn, G.
- CS Institut f. Anorg. u. Analyt. Chemie, TU Berlin, Berlin, 10623, Germany
- SO Book of Abstracts, 215th ACS National Meeting, Dallas, March 29-April 2 (1998), INOR-278 Publisher: American Chemical Society, Washington, D. C. CODEN: 65QTAA
- DT Conference; Meeting Abstract
- LA English
- AB Most catalytic olefin oligomerizations or polymns. procede via the hydride mechanism. An exception appears to be the selective trimerization of ethylene to 1-hexene with chromium catalysts—a reaction that may soon be an industrial process. However, the mechanism of this important transformation is still largly unclear. Jolly could support a possible mechanism with oxidative addn. of two olefins to a chromium complex giving a metallacyclopentane which inserts a third ethylene to give a metallacycloheptane before .beta.—hydride and reductive elimination regenerates the original complex and 1-hexene. We have found that N-substituted triazacyclohexane complexes of CrCl3 react with MAO to give a homogeneous catalyst for the selective trimerization of alpha-olefins. This allows the detailed study of the mechanism by observation of the reaction in the NMR or UV/Vis and by introducing olefinic N-substituents that participate in the initiation and trimerization.
- L3 ANSWER 2 OF 6 CAPLUS COPYRIGHT 2003 ACS
- AN 1997:784429 CAPLUS
- DN 127:354654
- TI The Chemistry of 1,3,5-Triazacyclohexane Complexes. 7. Synthesis and Characterization of the Cobalt(II) Methoxide Core {Co3(OMe)4}2+
- AU Koehn, Randolf D.; Haufe, Matthias; Kociok-Koehn, Gabriele; Filippou, Alexander C.
- CS Institut fuer Anorganische und Analytische Chemie, Technische Universitaet Berlin, Berlin, D-10623, Germany
- SO Inorganic Chemistry (1997), 36(26), 6064-6069 CODEN: INOCAJ; ISSN: 0020-1669
- PB American Chemical Society
- DT Journal
- LA English
- CC 78-7 (Inorganic Chemicals and Reactions) Section cross-reference(s): 75
- The authors report the synthesis and characterization of the 1st AB .eta.3-1,3,5-trimethyl-1,3,5-triazacyclohexane (Me3TAC) alkoxo complexes. Dehydration of [Co(H2O)6](X)2 with 2,2-dimethoxypropane or HC(OMe)3 and subsequent treatment with Me3TAC gave the methoxo cube fragment clusters  $[{Me3TAC}, Co}3(OMe)4](X)2(X = B(C6F5)4(2a) and B(m-C6H3(CF3)2)4(2b)).$ 2A crystallizes as a solvate in the triclinic space group P.hivin.1 (a 13.857(2), b 16.843(3), c 19.318(2) .ANG., .alpha. 79.23(2), .beta. 76.855(13), .gamma. 70.10(2).degree., Z = 2) and solvated 2b in the monoclinic space group P21 (a 12.710(3), b 23.251(3), c 18.702(4) .ANG., .beta. 101.16(2).degree., Z = 2). The cations contain a {Co3(OMe)4} cube fragment core with an .eta.3-Me3TAC bonded to each Co atom. Me3TAC is bonded unsym. due to H-bonding interactions with the anions. This broken symmetry can also be obsd. as solvent-, anion-, and H/D-isotope-dependent splitting of the signals in 1H and 2H NMR. A byproduct of the reaction is the protonated Me3TAC. The analogous monoacid adduct Me3TAC.cntdot.HCl was characterized by x-ray crystallog. (orthorhombic space group Pbca, a 11.0366(9), b 12.2986(6), c 13.9949(7) .ANG., Z = 8).
- ST crystal structure cobalt trimethyltriazacyclohexane methoxo trinuclear; structure cobalt trimethyltriazacyclohexane methoxo partial cubane; cobalt trimethyltriazacyclohexane methoxo partial cubane prepn; azacyclohexane cobalt methoxo partial cubane prepn; hydrogen bond cobalt trimethyltriazacyclohexane methoxo trinuclear
- IT Bond length

(cobalt-nitrogen; in cobalt trimethyltriazacyclohexane methoxo trinuclear partial cubane)

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IT
     NMR (nuclear magnetic resonance)
        (of cobalt trimethyltriazacyclohexane methoxo partial cubane)
IT
     Hydrogen bond
        (of cobalt trimethyltriazacyclohexane methoxo trinuclear partial
        cubane)
IT
     Crystal structure
     Molecular structure
        (of trimethyltriazacyclohexane hydrochloride and its cobalt methoxo
        trinuclear partial cubane)
IT
     Cluster compounds
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and crystal structure and hydrogen bonding of cobalt
        trimethyltriazacyclohexane methoxo partial cubane)
IT
     77-76-9, 2,2-Dimethoxypropane
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (for prepn. of cobalt trimethyltriazacyclohexane methoxo partial
        cubane)
     198197-24-9P
                    198197-26-1P
IT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (for prepn. of cobalt trimethyltriazacyclohexane methoxo partial
        cubane)
ΙT
     50-00-0, Formaldehyde, reactions
                                        74-89-5, Methylamine, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (for prepn. of trimethyltriazacyclohexane)
ΙT
     198197-29-4P
                   198197-33-0P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and crystal structure and hydrogen bonding of)
IT
     198197-22-7P
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (prepn. and crystal structure and reaction with cobalt aqua complex
        salts partial cubane)
     198197-30-7P
IT
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and mol. structure and hydrogen bonding of)
     ANSWER 3 OF 6 CAPLUS COPYRIGHT 2003 ACS
L3
ΑN
     1997:157895 CAPLUS
ΤТ
     Triazacyclohexane complexes of transition metals for olefin
     polymerization.
ΑU
     Koehn, R. D.; Haufe, M.; Seifert, G.; Kociok-Koehn, G.
CS
     Inst. f. Anorg. u. Analyt. Chemie, Technische Universitat Berlin, Berlin,
     D - 10623, Germany
SO
     Book of Abstracts, 213th ACS National-Meeting, San Francisco, April 13-17
     (1997), INOR-431 Publisher: American Chemical Society, Washington, D. C.
     CODEN: 64AOAA
     Conference; Meeting Abstract
DT
LΑ
     English
AB
     In the search for ancillary ligands enforcing unusual coordination
     geometries in complexes with potential catalytic activity, we investigated
     the coordination chem. of N-substituted 1,3,5-triazacyclohexanes (R3TAC).
    X-ray crystallog. on several complexes has shown that this ligand
     coordinates facially to a variety of transition metals with N-M-N bite
     angles of around 60.degree. and severely misdirected nitrogen lone pairs
     deviating from the direct N-M bond by 25-45.degree.. [(R3TAC)CrCl3]
     reacts with {MeAlO}.chi. (MAO) to solns. that polymerize ethylene. As a
     model for the active species we attempted to prep. complexes
     [(R3TAC)MR]n+. As a first structural model, [(R3TAC)ZnEt]+ was prepd.
     from ZnEt2 and R3TAC/PhNMe2H+. Several complexes (R3TAC)CrR'3 were prepd.
    However, we were unable to isolate the desired cationic Cr(III) complexes
    by protonolysis. Similar Cr(II) complexes were also investigated.
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- L3 ANSWER 4 OF 6 CAPLUS COPYRIGHT 2003 ACS
- AN 1996:584666 CAPLUS
- DN 125:276048
- TI The chemistry of 1,3,5-triazacyclohexane complexes. 5. Cationic zinc(II)

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alkyl complexes of N-alkylated 1,3,5-triazacyclohexanes and
     13-benzyl-1,5,9-triazatricyclo[7.3.1.05,13]tridecane
     Haufe, Matthias; Koehn, Randolf D.; Wiemann, Roman;
ΑU
     Seifert, Guido; Zeigan, Dieter
     Institut Fuer Anorganische und Analytische Chemie, Technische Universitaet
CS
     Berlin, Berlin, D-10623, Germany
     Journal of Organometallic Chemistry (1996), 520(1-2), 121-129
SO
     CODEN: JORCAI; ISSN: 0022-328X
PB
     Elsevier
DT
     Journal
LΑ
     English
     29-9 (Organometallic and Organometalloidal Compounds)
CC
     Section cross-reference(s): 75
GΙ
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Diethylzinc

173449-39-3

2547-66-2

RL: RCT (Reactant); RACT (Reactant or reagent)

Diethylzinc reacts with hydroperchlorates of N-alkylated AB 1,3,5-triazacyclohexanes (R3TAC; R = Me, PhCH2, iPr) and with the hydrotetrafluoroborate of 1,3,5-tris(p-fluorobenzyl)-1,3,5triazacyclohexane (FBz3TAC) to give the corresponding cationic Zn complexes [(R3TAC)Zn(Et)][X] (X = ClO4-, BF4-), e.g., I. Similar complexes were obtained from diethylzinc treated with [HNMe2Ph][BF4] or [HNMe2Ph][B(C6F5)4](Et2O) in the presence of R4TAC (R = PhCH2, p-FC6H4CH2, S-PhMeCH). The crystal structure of II, a product of the decompn. of I, was detd. The structures of [[(S-PhMeCH)3TAC]Zn(ET)][BF4] and [[(p-FC6H4CH2)3TAC]Zn(Et)][BF4] were estd. using nuclear Overhauser enhancement spectroscopy. Protonolysis of diethylzinc with [HNMe2Ph][BF4] in the presence of 13-benzyl-1,5,9-triazatricyclo[7.3.1.05,13]tridecane (BzTATC) yield [(BzTATC)Zn(Et)][BF4]. ST crystal structure triazacyclohexenium; mol structure triazacyclohexenium; zinc triazacyclohexane triazatricyclotridecane complex prepn; NOESY triazacyclohexane zinc complex; mol mechanics triazacyclohexane zinc complex IT Crystal structure Molecular structure (of triazacyclohexenium) IT 51720-36-6P RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (prepn. and crystal structure of) IT 118612-00-3P 182175-58-2P RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. of) IT 108-74-7 121-69-7, N,N-Dimethylaniline, reactions 557-20-0,

4520-86-9

10556-98-6

131968-96-2

```
(prepn. of cationic zinc(II) alkyl complexes of N-alkylated
        triazacyclohexanes and benzyltriazatricyclotridecane)
IT
     22533-76-2P
                   69737-15-1P
                                 182175-60-6P
                                                 182175-65-1P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (prepn. of cationic zinc(II) alkyl complexes of N-alkylated
        triazacyclohexanes and benzyltriazatricyclotridecane)
                                  182175-71-9P
TΤ
                    182175-70-8P
                                                  182175-72-0P
                                                                  182175-75-3P
     182175-77-5P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (prepn. of cationic zinc(II) alkyl complexes of N-alkylated
        triazacyclohexanes and benzyltriazatricyclotridecane)
     ANSWER 5 OF 6 CAPLUS COPYRIGHT 2003 ACS
L3
     1996:48776 CAPLUS
AN
     124:276702
DN
TI
     The chemistry of 1,3,5-triazacyclohexane complexes. Part 3. High yield
     synthesis of [Cr(N(SiMe3)2)3] and accurate structure determination by
     cocrystallization with Me6Si2
ΑU
     Koehn, Randolf D.; Kociok-Koehn, Gabriele; Haufe,
     Matthias
CS
     Inst. Anorg. Anal. Chem., Tech. Univ. Berlin, Berlin, D-10623, Germany
SO
     Chemische Berichte (1996), 129(1), 25-7
     CODEN: CHBEAM; ISSN: 0009-2940
PB
     VCH
DT
     Journal
LΑ
     English
CC
     78-5 (Inorganic Chemicals and Reactions)
     Section cross-reference(s): 75
AB
     The reaction of the [CrLCl3] (L = N, N', N''-trimethyl-1, 3, 5-
     triazacyclohexane) with LiN(SiMe3)2 or NaN(SiMe3)2 in petroleum ether
     yielded nearly quant. [Cr{N(SiMe3)2}3] (I) with loss of L. I was crystd.
     from Me6Si2 as [Cr{N(SiMe3)2}3].cntdot.(Me6Si2)0.5 which allowed the
     refinement of the x-ray crystal structure in the trigonal space group P31c
     (a 16.012, c 8.4796 .ANG., V = 1882.8 .ANG.3, Z = 2) without severe
     disorder.
ST
     triazacyclohexane chromium complex reaction silylamine; chromium
     silylamine complex prepn; crystal structure chromium silylamine complex;
     mol structure chromium trimethylsilylamine complex
ŦΤ
     Crystal structure
     Molecular structure
        (of chromium bis(trimethylsilyl)amide)
IT
     175285-80-0P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (crystal structure; prepn. of bis(trimethylsilylamine) chromium from
        triazacyclohexane chromium and bis(trimethylsilylamine) lithium)
IT
     175285-79-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (des prepn. of bis(trimethylsilylamine) chromium from
        triazacyclohexanechromium and bis(trimethylsilylamine) lithium)
TT
     4039-32-1, Lithium Bis(trimethylsilyl)amide
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (prepn. of bis(trimethylsilylamine) chromium from triazacyclohexane
        chromium and bis(trimethylsilylamine) lithium)
L3
     ANSWER 6 OF 6 CAPLUS COPYRIGHT 2003 ACS
AN
     1995:847433 CAPLUS
DN
     124:56199
     .eta.3-1,3,5-Triazacyclohexane complexes of tribenzylchromium(III)
TI
ΑU
     Koehn, Randolf D.; Kociok-Koehn, Gabriele; Haufe,
     Matthias
CS
     Institut fuer Anorganische und Analytische Chemie, Technische Universitaet
     Berlin, Strasse des 17, Juni 135, Berlin, D-10623, Germany
SO
     Journal of Organometallic Chemistry (1995), 501(1-2), 303-7
     CODEN: JORCAI; ISSN: 0022-328X
PB
     Elsevier
DT
     Journal
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LA
     English
     29-11 (Organometallic and Organometalloidal Compounds)
CC
     Section cross-reference(s): 75
AΒ
     1,3,5-Tribenzyl-1,3,5-triazacyclohexane ('Bz3TAC') and
     1,3,5-tricyclohexyl-1,3,5-triazacyclohexane ('Cy3TAC') react with
     CrCl3(THF)3 to the corresponding complexes (Bz3TAC)CrCl3 (1a) and
     (Cy3TAC)CrCl3 (1b). These react with benzylsodium in toluene to the
     tribenzylchromium(III) complexes (Bz3TAC)Cr(CH2Ph)3 (2a) and
     (Cy3TAC)Cr(CH2Ph)3 (2b). The crystal and mol. structure of
     2b.cntdot.1.5toluene was detd. by x-ray diffraction.
     crystal structure chromium benzyl triazacyclohexane complex; mol structure
ST
     chromium benzyl triazacyclohexane complex
IT
     Crystal structure
     Molecular structure
        (of chromium benzyl triazacyclohexane complex)
     Crystal field theory and Ligand field theory
IT
        (of chromium triazacyclohexane complexes)
IT
     Energy level splitting
        (crystal-field, of chromium triazacyclohexane complexes)
     1121-53-5, Benzylsodium
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (for prepn. of chromium benzyl triazacyclohexane complexes)
ΙT
     2547-66-2, 1,3,5-Tribenzyl-1,3,5-triazacyclohexane
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (for prepn. of chromium complexes)
     6281-14-7P, 1,3,5-Tricyclohexyl-1,3,5-triazacyclohexane
IT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (for prepn. of chromium complexes)
     10170-68-0, Trichlorotris (tetrahydrofuran) chromium
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (for prepn. of triazacyclohexane complexes)
TΤ
     108-91-8, Cyclohexylamine, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (for prepn. of tricyclohexyltriazacyclohexane)
IT
     172166-81-3P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and crystal structure of)
IT
     172166-84-6P, Tris(benzyl)(1,3,5-tribenzyl-1,3,5-
     triazacyclohexane) chromium
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. and ligand-field parameters)
     172166-82-4P, Trichloro(1,3,5-tribenzyl-1,3,5-triazacyclohexane)chromium
IT
     172166-83-5P, Trichloro(1,3,5-tricyclohexyl-1,3,5-
     triazacyclohexane) chromium
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (prepn., ligand-field parameters and reaction with benzylsodium)
     172166-80-2P, Tris(benzyl)(1,3,5-tricyclohexyl-1,3,5-
ΙT
     triazacyclohexane) chromium
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
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(prepn., mol. structure and ligand-field parameters)